

# ENZYMES

Nafith Abu Tarboush

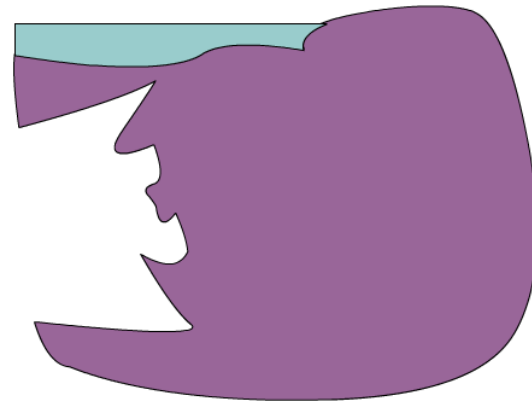
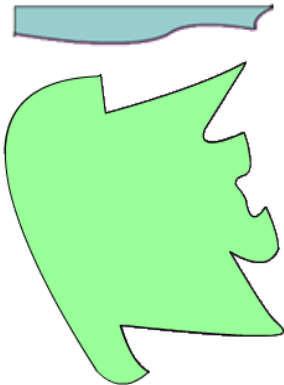
DDS, MSc, PhD

[natarboush@ju.edu.jo](mailto:natarboush@ju.edu.jo)

[www.facebook.com/natarboush](https://www.facebook.com/natarboush)

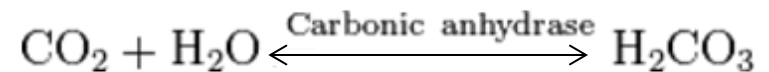
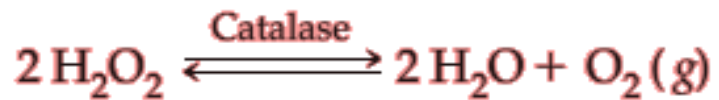
# General properties of proteins

- The function of nearly all proteins depends on their ability to bind other molecules (ligands)
- Two properties of a protein characterize its interaction with ligands:
  - **Affinity: the strength of binding between a protein and other molecules**
  - **Specificity: the ability of a protein to bind one molecule in preference to other molecules**



# The Biological Catalysts; Enzymes

- What are enzymes? (specialized proteins, small amounts, acceleration, no change). Ribozymes are the exception
- Enzymes are the most efficient catalysts known
  - Usually in the range of  $10^6$  to  $10^{14}$
  - Non-enzymatic catalysts ( $10^2$  to  $10^4$ )
  - The actions of enzymes are fine-tuned by regulatory processes
- Examples: catalase ( $10^8$ ) & carbonic anhydrase ( $10^7$ )

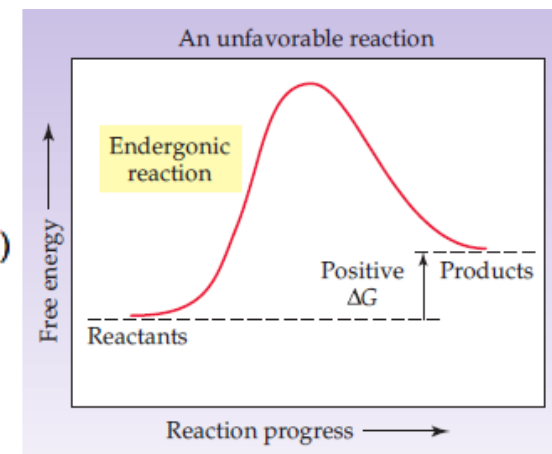
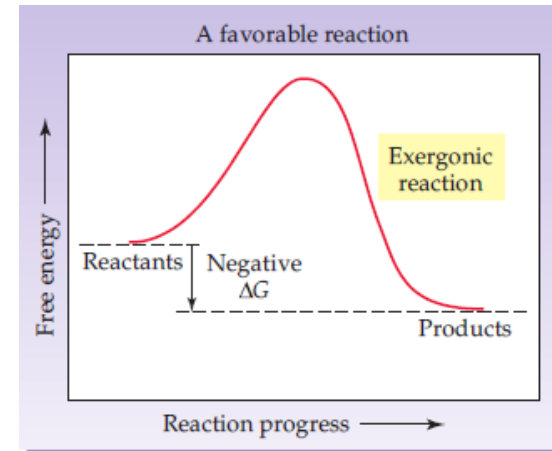
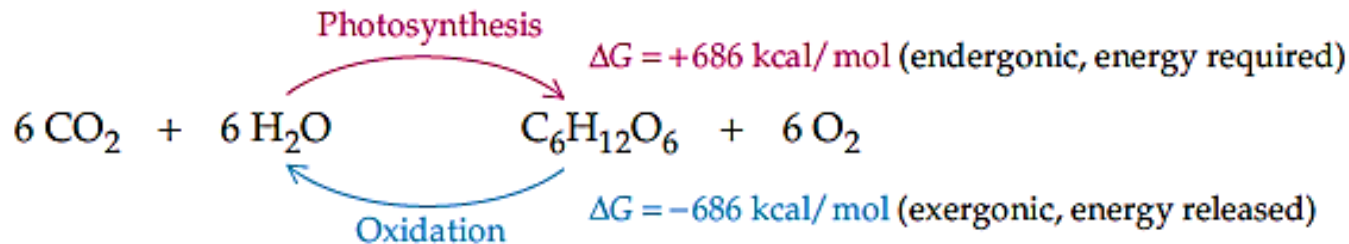


Reaction Conditions	Activation Free Energy		Relative Rate
	$\text{kJmol}^{-1}$	$\text{kcal mol}^{-1}$	
No catalyst	75.2	18.0	1
Platinum surface	48.9	11.7	$2.77 \times 10^4$
Catalase	23.0	5.5	$6.51 \times 10^8$



# Energy & Biochemical Reactions

- $\Delta G = \Delta H - T\Delta S$
- Spontaneous vs. non-spontaneous, favorable vs. non-favorable, exergonic vs. endergonic, exothermic vs. endothermic, switch of signs
- $\Delta G$ ,  $\Delta G^\circ$
- Biochemical pathways; storage (endergonic) & release (exergonic)
- Kinetics (rate) vs. Thermodynamics (favorability)



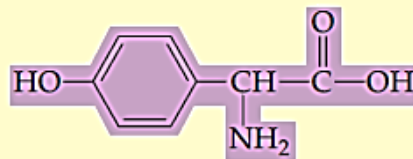
# Are enzymes important?

- In the human body, almost every metabolic process involve the use of enzymes

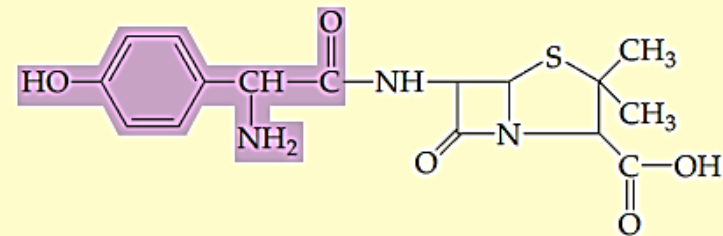


Sucrose (table sugar), yeast enzyme breaks sucrose into its two smaller sugar

crushed leaves are exposed to the oxygen in air, a polyphenoloxidase breaks up polyphenols into tannins which impart the darker color and characteristic flavors



*p*-Hydroxyphenylglycine



Amoxicillin

# How to express an enzymatic reaction?

- In enzymatic reactions, reactants are known as substrates
- We can simply express an enzymatic reaction using this formula



Or



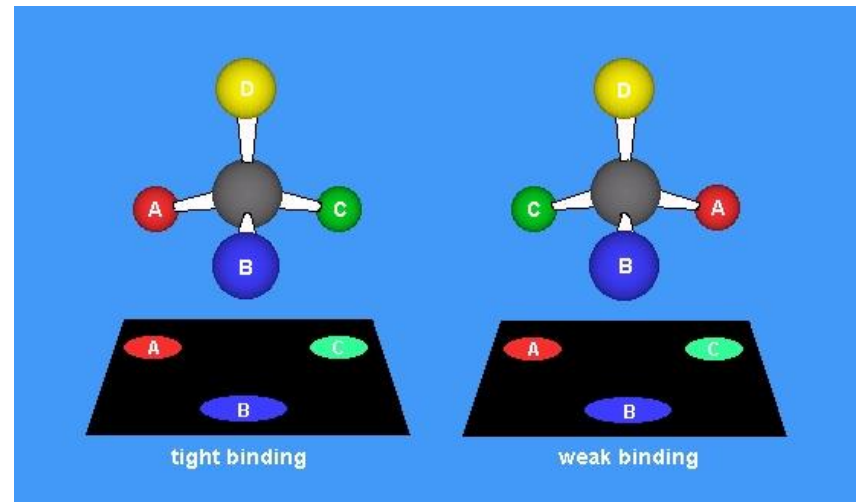
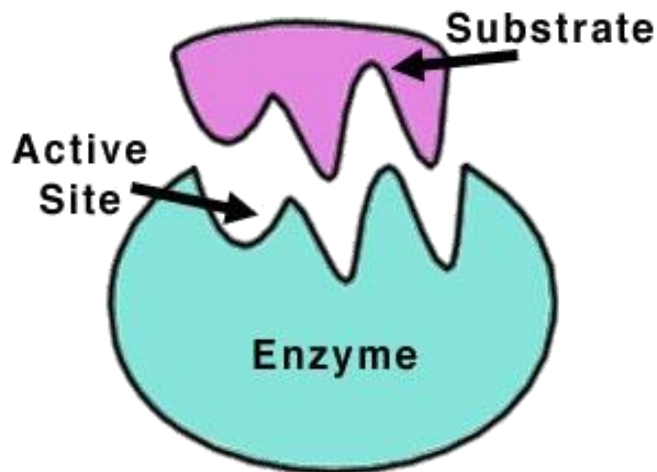
where E is the free enzyme; S is the free substrate, ES is the enzyme-substrate complex; P is the product of the reaction; and EP is the enzyme-product complex before the product is released

# Active sites of enzymes

- A specific three-dimensional shape which includes a region where the biochemical reaction takes place
- Contains a specialized amino acid sequence that facilitates the reaction
- Within the active site are two sub-sites, the binding site and the catalytic site, The binding & catalytic site may be the same
- Binding site: binds substrate through ionic, H-bonding or other electrostatic forces
- Catalytic site: contains the catalytic groups

# Features of active site

- Active sites; structures that look like canals, clefts or crevices
- Water is usually excluded after binding unless it participates in the reaction
- Substrates are bound to enzymes by multiple weak attractions (electrostatic, hydrogen, van der Waals, & hydrophobic)
- Binding occurs at least at three points (chirality)





# Features of active site

- Forms by groups from different parts of the amino acid sequence usually forming a domain made of multiple secondary structures
- Takes up a relatively small part of the total volume
- The “extra” amino acids help create the three-dimensional active site & in many enzymes, may create regulatory sites

## ALLOSTERIC ENZYME

